## BIOLOGICALLY ACTIVE SUBSTANCES FROM Cacalia hastata LEAVES. 5. COUMARINS AND TRITERPENES

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In continuation of research on the chemical composition of *Cacalia hastata* L. leaves, a batch (1.0 kg) was processed in a Soxhlet apparatus successively by  $C_6H_{14}$ ,  $CHCl_3$ , and EtOAc. After saponification (10% KOH/EtOH), the  $CHCl_3$  extract was diluted in water (1:5). The resulting solid was filtered off and chromatographed over silica gel ( $CHCl_3$ :EtOH,  $100:0 \rightarrow 90:10$ ) with TLC monitoring ( $C_6H_6$ :MeOH, 4:1;  $C_6H_{14}$ : $C_6H_6$ : $CHCl_3$ :MeOH, 5:4:2:1) to isolate two compounds 1 and 2.

The aqueous phase after saponification was acidified and extracted with  $CHCl_3$ . The extract was chromatographed over silica gel ( $CHCl_3$ :EtOH,  $100:0 \rightarrow 95:5$ ) with paper chromatographic (formamide: $CHCl_3$ , system I) and TLC ( $CHCl_3$ :EtOH, 19:1, system II) monitoring. The isolated compounds were rechromatographed in a free layer of silica gel (40/100, system I) and recrystallized from MeOH to afford three compounds 3-5.

Qualitative reactions established that 1 and 2 were triterpenes; 3-5, coumarins. The isolated compounds were identified using UV spectroscopy, chromatographic mobility, melting and mixed melting points, chemical transformations, and optical rotations.

Compound 1, yield 0.05%, mp 139-140°C,  $[\alpha]_D^{20}$  -36° (c 0.5, CHCl<sub>3</sub>) [1], identified as  $\beta$ -sitosterol.

Compound **2**, oleanolic acid [2], yield 0.12%, pale yellow needlelike crystals, mp 308.0-309.8°C,  $[\alpha]_D^{20}$  +78° (c 0.5, CHCl<sub>3</sub>), UV spectrum (H<sub>2</sub>SO<sub>4</sub>,  $\lambda_{max}$ , nm) 310 [1]. Acetylation of **2** produced the acetate with mp 257-258°C.

Compounds 3-5 were reduced with hydriodic acid [3] to give a compound with mp 66-68°C, UV spectrum (EtOH,  $\lambda_{max}$ , nm) 272, 328, that was characterized as coumarin [4].

Compound 3, yield 0.015%, mp 202-203°C (MeOH), TLC  $R_f$  0.55 (system I), 0.52 (system II), UV spectrum (EtOH,  $\lambda_{\rm max}$ , nm) 231, 295, 340, scopoletin [4].

Compound **4**, yield 0.009%, mp 230-232°C (EtOH), TLC  $R_f$  0.18 (system I), 0.041 (system II), UV spectrum (EtOH,  $\lambda_{\text{max}}$ , nm) 228, 256, 330, umbelliferone [5]. The structure of **4** was confirmed by direct synthesis from malic acid and resorcinol by a Pechmann reaction [6].

Compound **5**, yield 0.001%, mp 269-270°C (EtOH), TLC  $R_f$  0.02 (system I), 0.15 (system II), UV spectrum (EtOH,  $\lambda_{\text{max}}$ , nm) 262, 305, 356, esculetin [5].

Coumarins in C. hastata leaves were determined quantitatively using chromatography and spectrophotometry and the specific extinction coefficients E (1 cm, 1%) [7]; triterpenes calculated for oleanolic acid, by spectrophotometry using a halochromic reaction with conc.  $H_2SO_4$  [8]. It was found that the contents of scopoletin, esculetin, and triterpenes in C. hastata leaves increased during mass flowering, reaching values of 0.031, 0.002, and 0.92%, respectively, and slightly decreased by autumn (Table 1). The exception was umbelliferone, the concentration of which increased toward the end of vegetation.

Triterpenes in representatives of the genus *Cacalia* L. have been observed in *C. aconitifolia* (oleanolic acid acetate) [2] and *C. kamtschatica* ( $\beta$ -amurin,  $\beta$ -sitosterol, stigmasterol, campesterol) [9, 10]. Coumarins in *C. hastata* have not been previously investigated. All compounds were isolated from *C. hastata* leaves for the first time.

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TABLE 1. Accumulation Dynamics of Coumarins and Triterpenes in C. hastata L. Leaves, %

Substance	Start of vegetation (June 1)	Vegetation (June 11)	Vegetation (June 22)	Budding (July 28)	Mass flowering (August 15)	Fruiting (August 28)
Scopoletin	0.010±0.001	0.010±0.002	0.013±0.002	0.028±0.004	0.031±0.005	0.012±0.005
Umbelliferone	$0.005\pm0.001$	$0.006\pm0.001$	$0.015\pm0.001$	$0.017 \pm 0.001$	$0.020\pm0.002$	$0.028\pm0.003$
Esculetin	-	-	0.001	0.001	0.002	0.001
Triterpenes	$0.31\pm0.03$	$0.44\pm0.04$	$0.74\pm0.09$	$0.86\pm0.08$	$0.92\pm0.11$	$0.82\pm0.09$

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